

Molecular layers in the dust formation zone of AGB stars

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Seminar

Nice 4th of June 2015

Overview

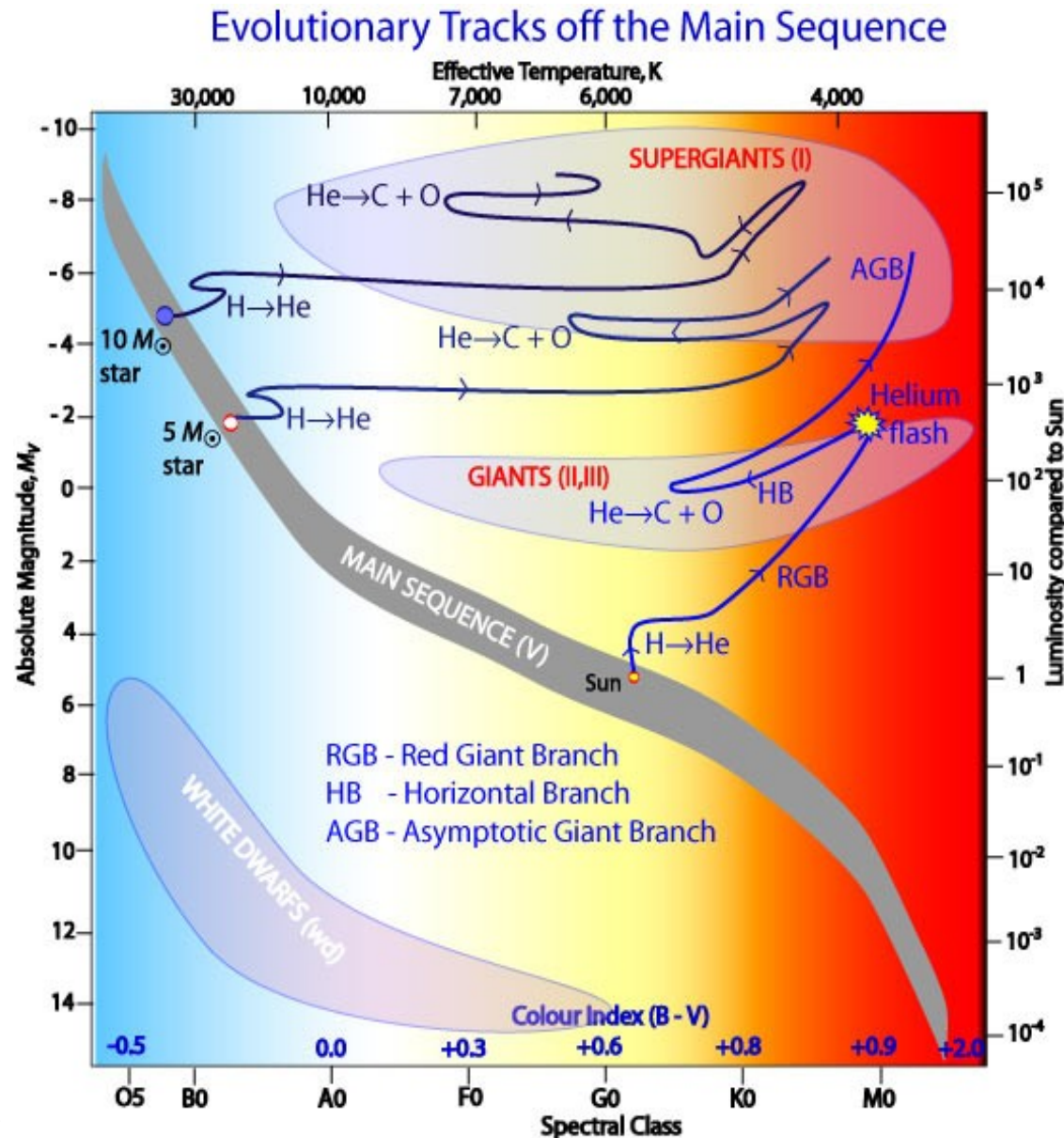
- AGB stars: types & evolution
- The inner wind of AGB stars
- Stationary wind vs. Dynamic pulsations
- Chemical model for the gas phase, cluster formation routes and dust grains
- Results on molecules & dust clusters
- Dust condensation & grain size distributions
- Outlook & Conclusions

AGB stars

Late stage evolution of low- and intermediate mass stars ($M_{*}^{\text{ZAMS}} < 8 M_{\odot}$)

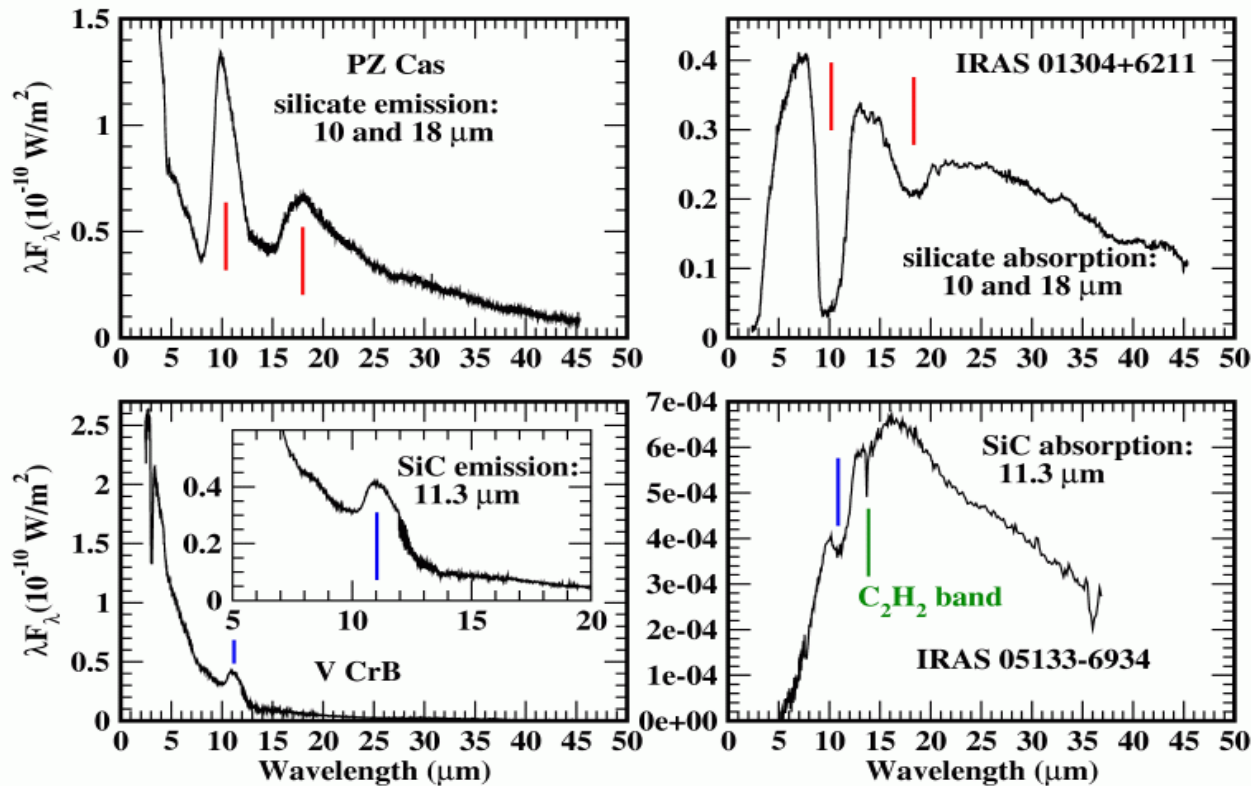
Cool photospheres ($T = 2000 - 3000 \text{ K}$) and high luminosities ($L \sim 10^3 - 10^4 L_{\odot}$)

High mass loss rates ($10^{-7} - 10^{-4} M_{\odot}/\text{yr}$) due to the presence of dust



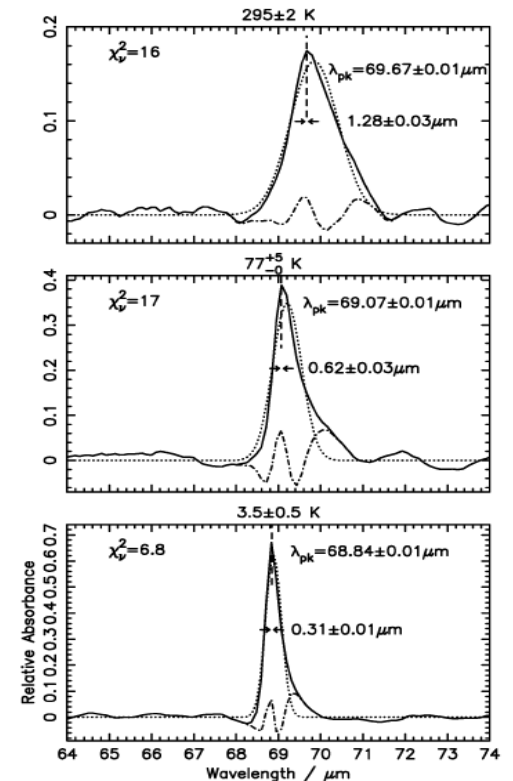
Evidence for dust

Spectral dust features (Silicates & SiC):



Homepage StSci, K. Volk

Forsterite (Mg_2SiO_4) in the lab (69 μm):



Bowey 2013

In addition: MgS at 30 μm

PolyAromatic Hydrocarbons (PAHs) at 3.3, 6.2, 7.7, 8.8, 12.7 and 16.4 μm \rightarrow (hydrogenated) amorphous carbon

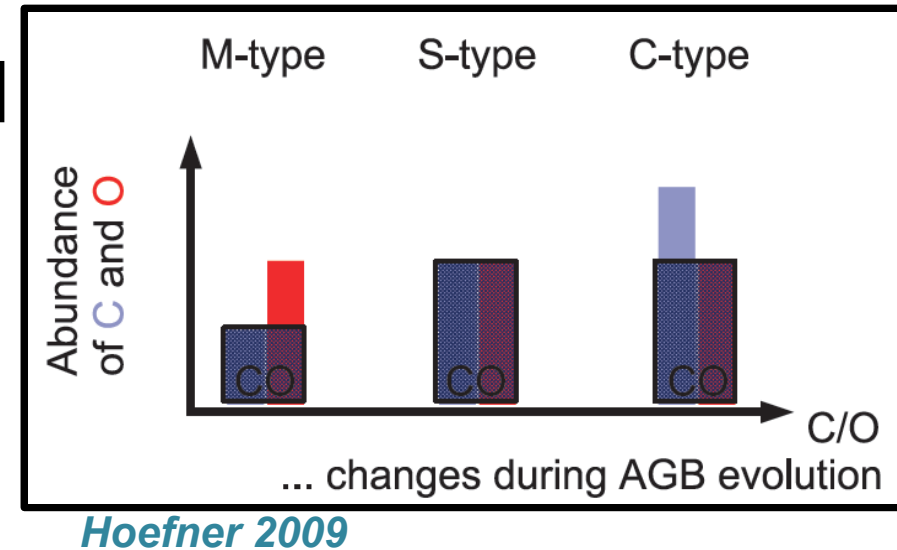
Chemical types of AGB stars

M-type: oxygen-rich, $C/O < 1$

S-type: $C/O \sim 1$

C-type: carbon-rich, $C/O > 1$

... at the photosphere



Low mass: 3rd dredge up mixes carbon to the photosphere → C/O increases

Intermediate mass: Hot bottom burning converts C in N and other CNO products

→ C/O decreases

C/O ratio: molecules

Thermodynamic equilibrium (TE) predicts the presence of:

CO, C, HCN, CS, C₂H₂, CH, CN in the C-rich case

Tsuij 1973

CO, H₂O, SiO, OH, TiO in the O-rich case

→ $E_{\text{bind}}(\text{CO}) = 11.2 \text{ eV}$ → locks up lesser abundant element

BUT observed are also:

C-bearing molecules in O-rich AGBs:

HCN, CS, OCS, CN, CO₂ *Deguchi 1985, Lindqvist 1988, Omont 1993, Bujarrabal 1993, Justtanont 1998, Decin 2010*

O-bearing species in carbon stars:

H₂O, OH, SiO *Melnick 2011, Hasegawa 2006, Ford 2004, Schoeier 2006*

TE cannot account for these observed molecules

=> Kinetic description is necessary for the chemistry

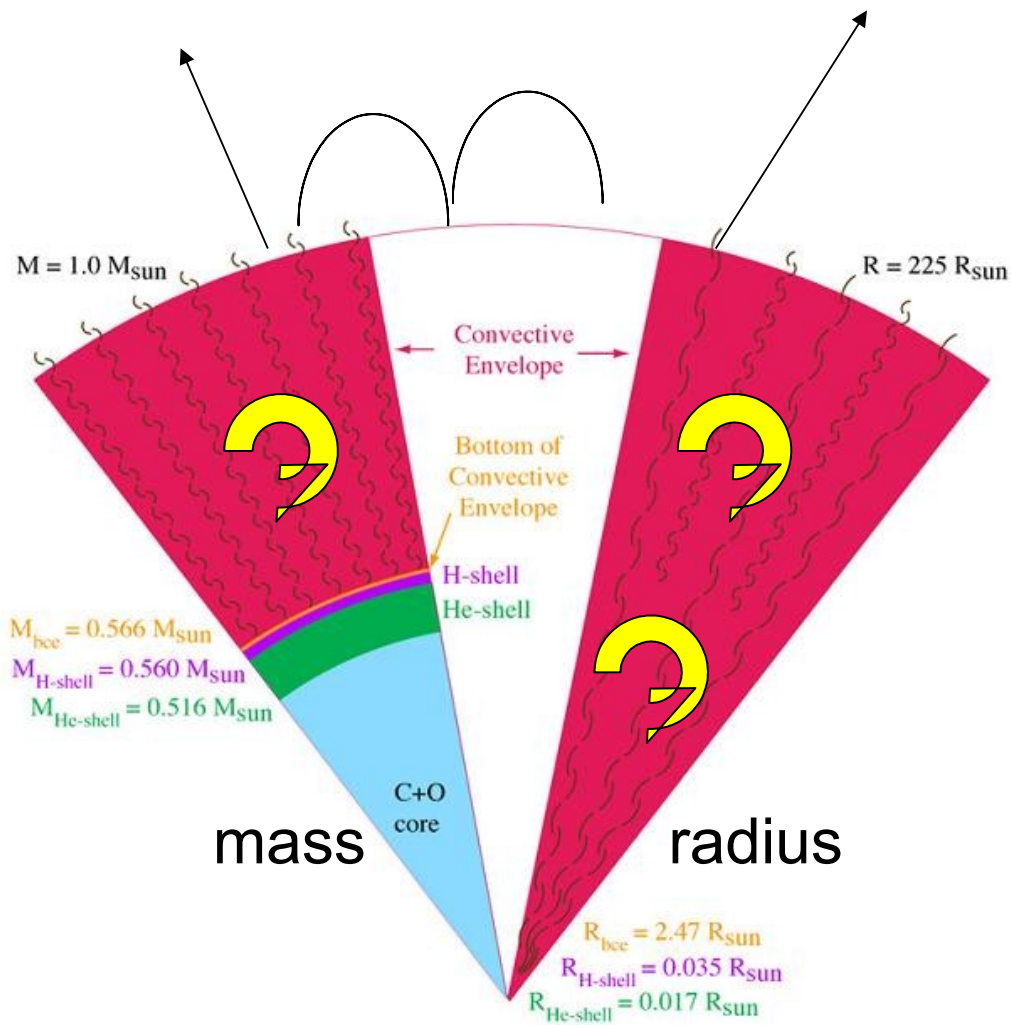
AGB star structure

Stellar wind driven by pulsations and dust grains

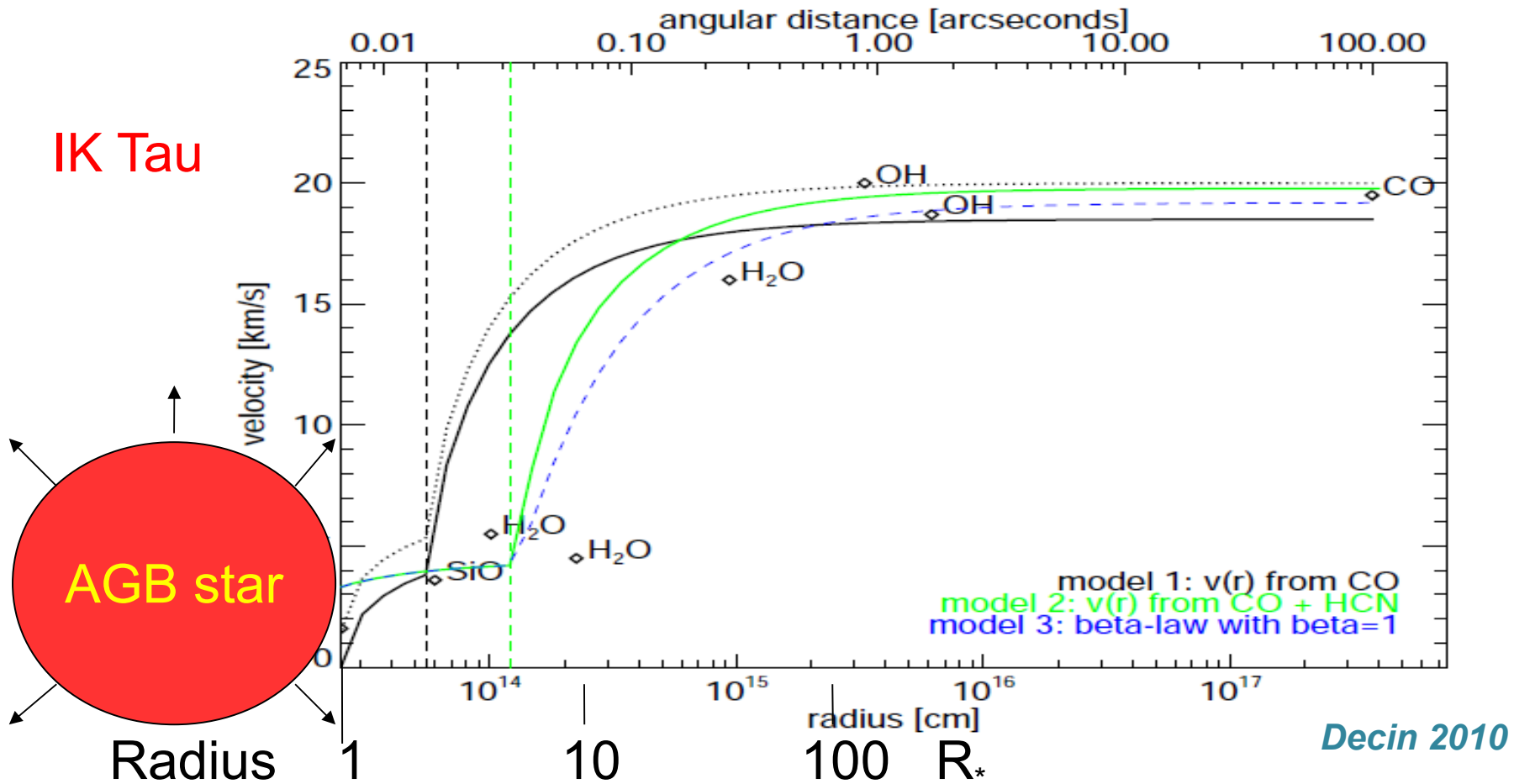
Large convective envelope

Energy generation proceeds in thin burning shells

Degenerate (non-burning) core consisting of carbon and oxygen



The inner wind of AGBs



**Pulsating
Photosphere**

T = 2200 K
N = 10^{14} cm^{-3}

CO, H₂O
SiO

**Inner envelope
Dust formation**

500-42'000 K
 $10^6 - 10^{16} \text{ cm}^{-3}$

HCN, CS, CO₂
SiS, PN, AlOH

Fully accelerated wind

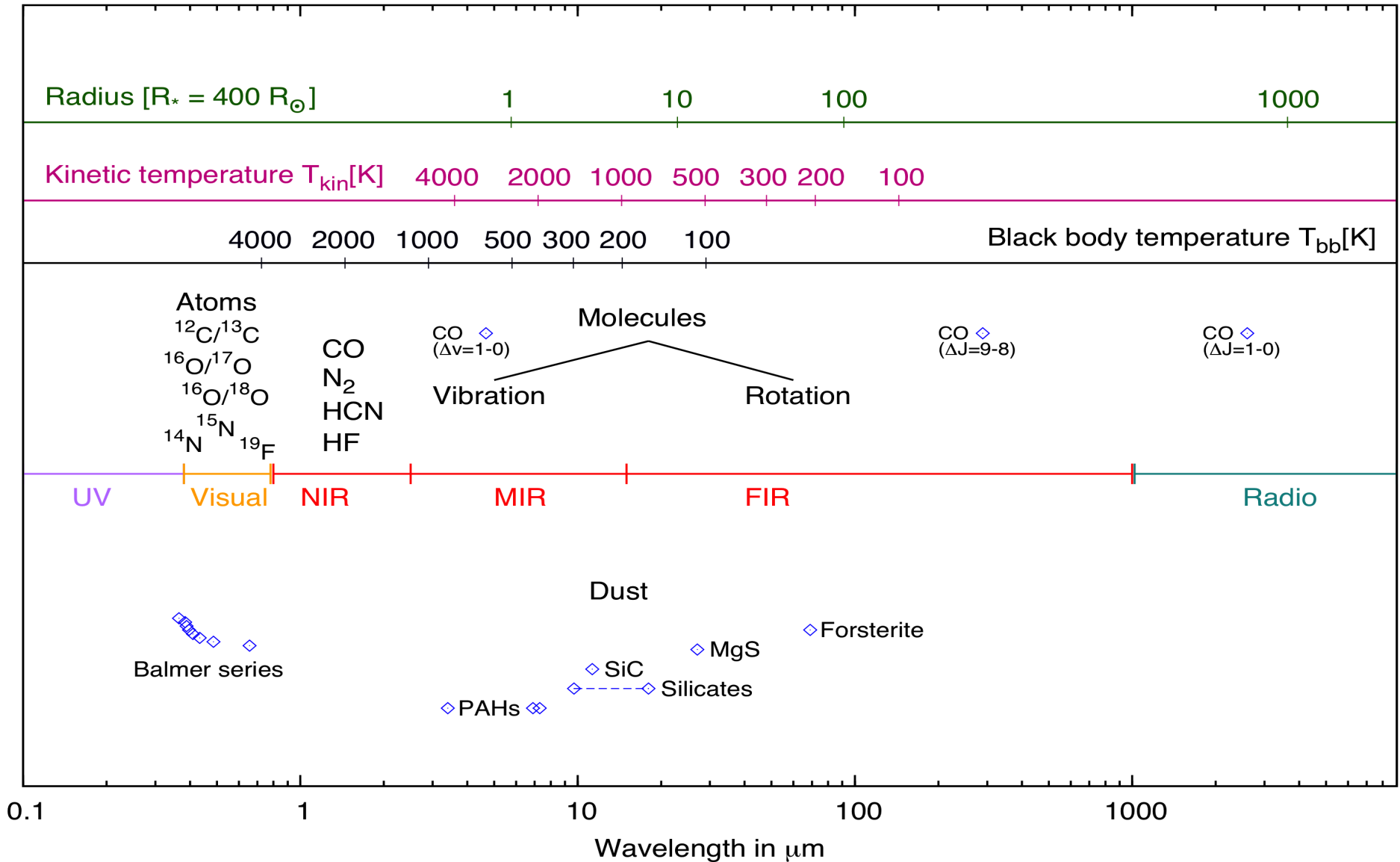
100 K
 10^6 cm^{-3}

NH₃, SO₂
H₂S

daughter species produced by
photo reactions of mother species

Interstellar
UV radiation

Energy scale in the wind



Timescales for AGB stars

Main sequence: $\sim 10^7 - 10^{10}$ years

Time on the (TP) AGB: $\sim 10^6$ years

Time between dredge-up (mixing) episodes:
 $\sim 10^4$ years *Maercker 2009*

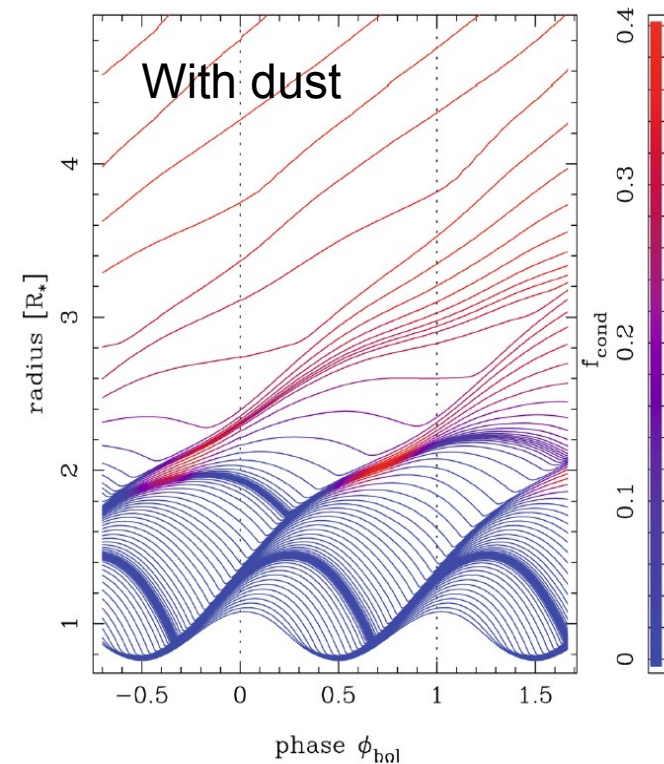
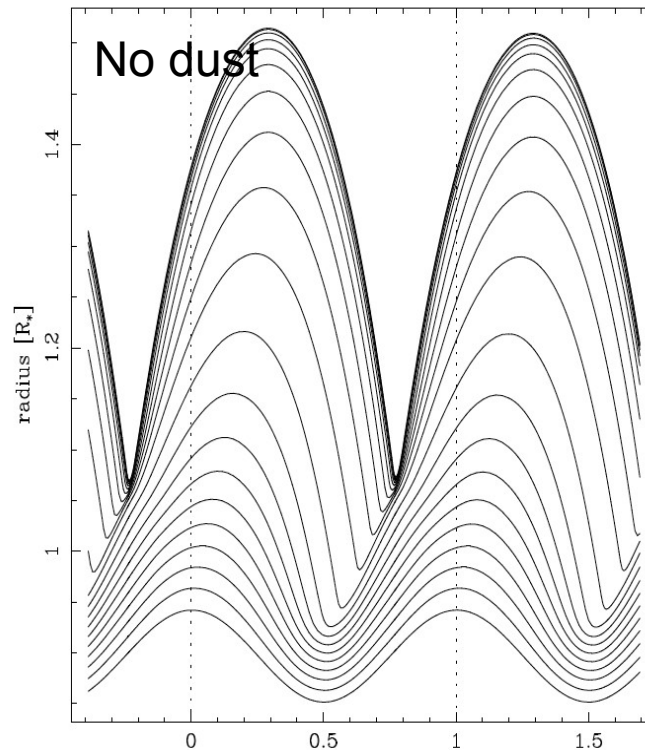
Timescale in the wind:

- Expansion timescale: $R_{\text{env}} / V_{\text{exp}} \sim$
- Time between pulsations: ~ 1 year
- Chemical reactions: $1/k \sim$ ms to hours
- Dust growth: coagulation & surface growth

Physics of the inner wind

Periodic shocks cross the stellar atmosphere

→ ambient gas is compressed, heated and accelerated



Nowotny 2010

- formation of dense and warm gas layers gravitationally bound to the star
- favourable conditions for molecule, cluster & dust grain formation [Cherchneff 1996/2006/11/12](#), [Willacy & Cherchneff 1998](#), [Duari 1999](#)
- ***Pulsations & dust are necessary for mass loss***

Periodic pulsation model for a galactic Mira star: IK Tau

Pre-shock profiles:

Temperature

number density

Scale height

$$T(r) = T_* \left(\frac{r}{R_*} \right)^{-0.6}$$

$$n(r) \propto n_* \exp \left(\frac{-r}{H} \right)$$

$$H = \frac{k_B T}{\mu m_H g} = \frac{k_B T R^2}{\mu m_H G M}$$

Willson & Bowen 1984, Cherchneff 1992

Post-shock profiles:

- Rankine-Hugoniot jump conditions
- Gas excursions on ballistic trajectories

Based on mass,
momentum, and
energy
conservation

$$\frac{d\rho}{dt} \equiv \left(\frac{\partial}{\partial t} + v \frac{\partial}{\partial z} \right) \rho = -\rho(\nabla \cdot \mathbf{v}) ,$$

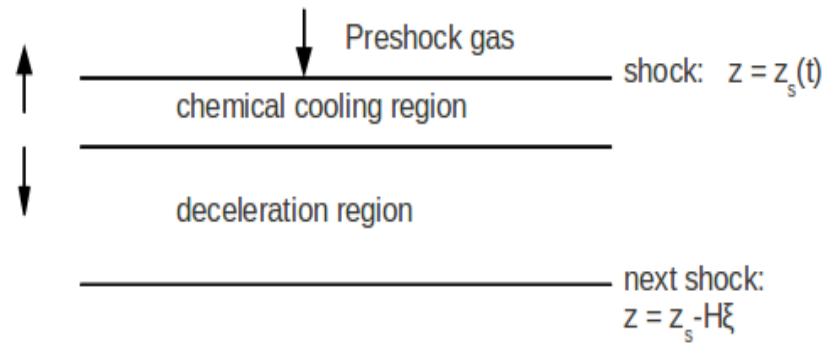
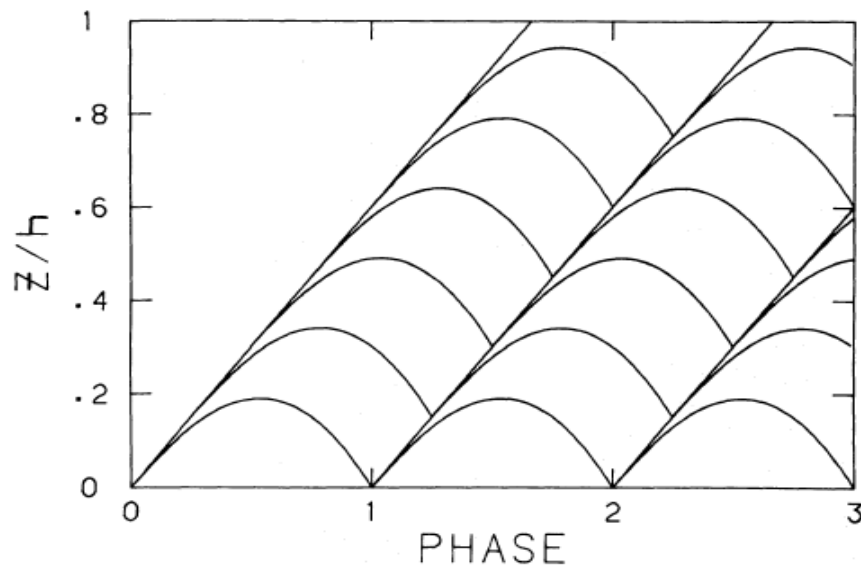
$$\frac{dv}{dt} + \frac{1}{\rho} \frac{\partial p}{\partial z} = -g ,$$

$$\frac{dp}{dt} + \gamma p(\nabla \cdot \mathbf{v}) = 0 .$$

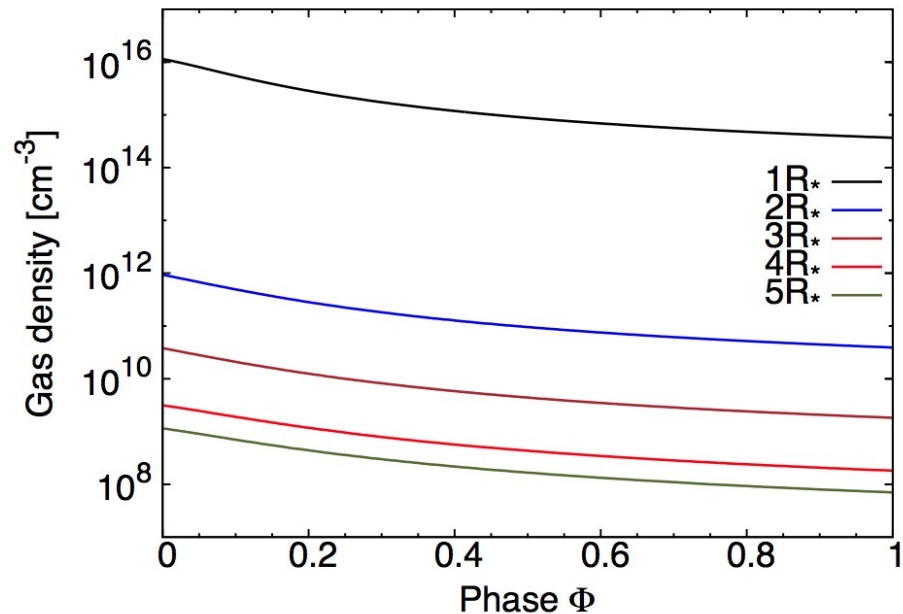
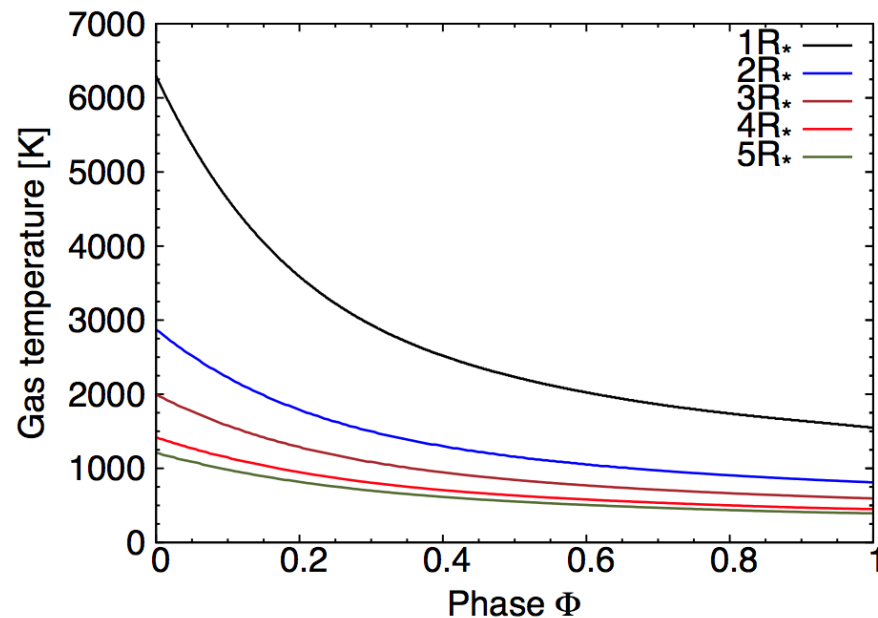
Parameters for IK Tau

Parameter	Value
T_*	2200 K
M_*	$1 M_\odot$
R_*	2.5×10^{13} cm
v_s	25 – 32 kms ⁻¹
P	470 days
$n(1R_*)$	3.6×10^{14} cm ⁻³
r_s	$1 R_*$
C/O	0.75
\dot{M}	$(0.4 - 3) \times 10^{-5} M_\odot/\text{yr}$
Ψ	1.9×10^{-2}

Periodic pulsation model for Miras



Fox & Wood 1985
Bertschinger & Chevalier 1985
Cherchneff 1996
Willacy & Cherchneff 1998
Duari et al. 1999
Cherchneff 2006, 2011 & 2012



Gas-phase chemistry

Molecular detections in inner oxygen-rich AGB winds:

H₂O, OH, SiO, SiS, NaCl, **CO, CO₂, HCN, CS**, SO, SO₂, NH₃

PN, PO *Milam 2007, Justtanont 1998, Decin 2010, Justtanont 2012, Menten 2010, De Beck 2013*

Chemical network contains termolecular & bimolecular (neutral-neutral, collisional fragmentation, radiative association) processes – no ions

Reaction type	Reaction	formulation	Gas conditions
<i>Unimolecular</i>			
Thermal decomposition	AB	$\rightarrow A + B$	High T
<i>Bimolecular</i>			
Neutral-exchange	AB + C	$\rightarrow A + CB$	T dependent
Collisional dissociation	AB + M	$\rightarrow A + B + M$	High T
Radiative association	A + B	$\rightarrow AB + \gamma$	T independent
<i>Termolecular</i>			
Termolecular formation	A + B + M	$\rightarrow AB + M$	(Very) high n

17 elements
105 molecules
426 reactions

GOAL: Reproduce these molecules in the observed amounts

Chemical network

Bimolecular reaction: $A + B \rightarrow C + D$

Change in number density of species C:

$$\frac{dn(C)}{dt} = k_{AB} n(A) n(B) \quad k_{AB} = \alpha \left(\frac{T}{298 \text{ K}} \right)^\beta \exp \left(\frac{-E_a}{RT} \right)$$

Set of reactions :

Arrhenius reaction rate

$$\frac{dn_i}{dt} = \sum_{j,k} k_{jk} n_j n_k + \sum_{j,k} k_{jkM} n_j n_k n_M - n_i \sum_l k_{il} n_l - n_i \sum_n k_n$$

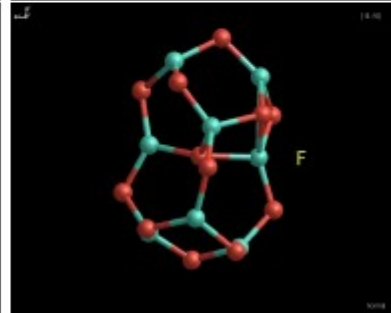
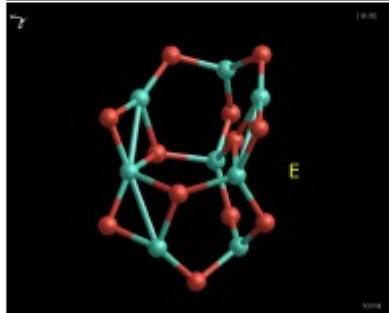
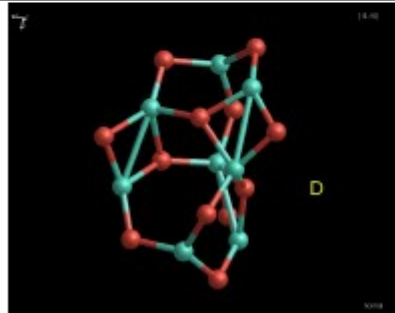
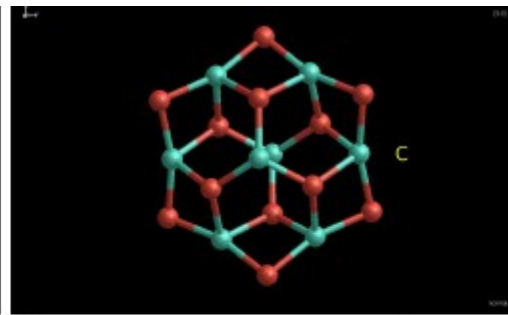
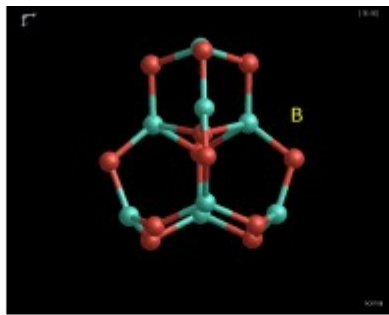
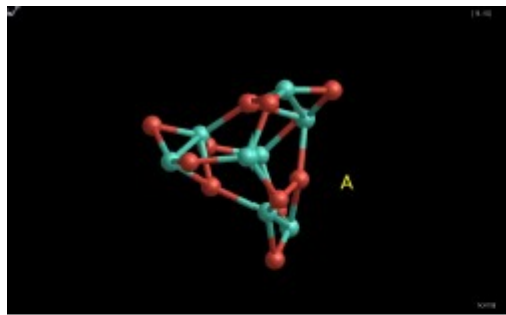
→ set of stiff, coupled, non-linear ordinary differential equations (ODEs)

→ ODEs subject to varying temperature and densities

Cluster nucleation routes

Chemical network involves formation pathways to dimers of alumina (Al_2O_3) & forsterite (Mg_2SiO_4), enstatite (MgSiO_3) of metal oxides (SiO , MgO , FeO , TiO), and pure metal clusters (Fe , Al , Si)

- Structure determination by a semi-classical Monte-Carlo-based candidate search & subsequent quantum Density Functional Theory (DFT) calculations
- DFT analysis yields thermochemical properties, which indicate trends for reaction mechanisms and kinetics



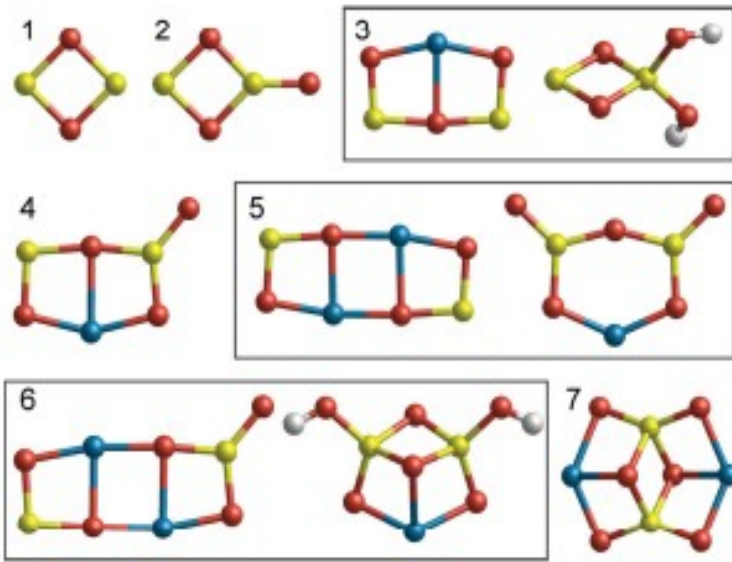
$$\Delta G = \Delta H - T\Delta S$$



Energetically
favoured
alumina
tetramers

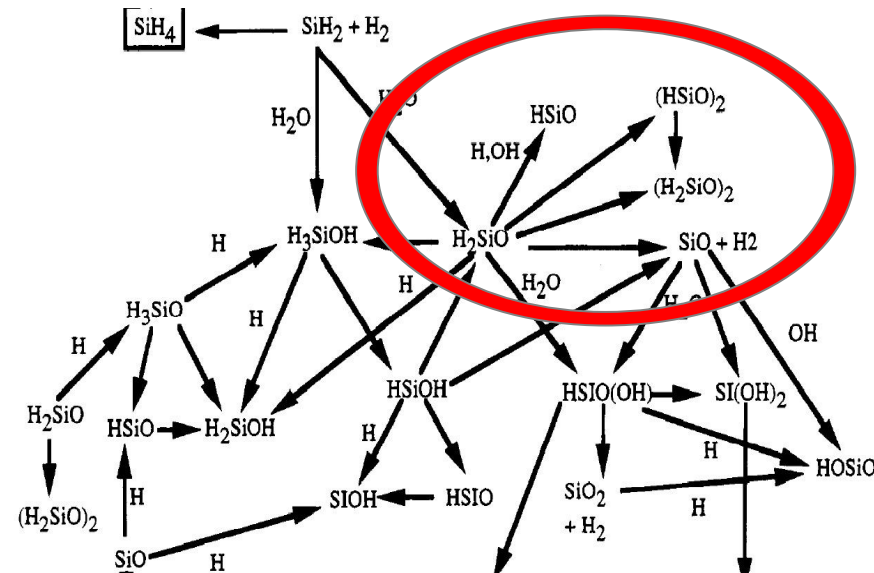
Cluster nucleation routes

Silicates: forsterite dimers $\text{Mg}_4\text{Si}_2\text{O}_8$



Goumans & Bromley 2012

Enstatite formation pathway



Zachariah & Tsang 1995

Formation of hydrogenated silicon oxides
in silane-rich flame

- SiO dimerisation too slow to start silicate nucleation
- Nucleation proceeds via HSiO , $\text{H}_2\text{Si}_2\text{O}_2$ & $\text{H}_2\text{Si}_2\text{O}_3$ formation
- Growth via successive oxidation & Mg inclusion steps
- Efficient mechanism to synthesise silicate dimers (enstatite and forsterite) between $\sim 4 R_*$ and $6 R_*$

Dust grain condensation

Formalism based on Brownian thermal motion, which accounts for the scattering, fragmentation, and coagulation of the grains *Plane 2013, Sarangi & Cherchneff 2014*

→ Grains size distributions are derived for silicates of forsterite and enstatite stoichiometry, and alumina.

→ Grains are assumed to be spherical, volume conserving and stable to (stellar) radiation

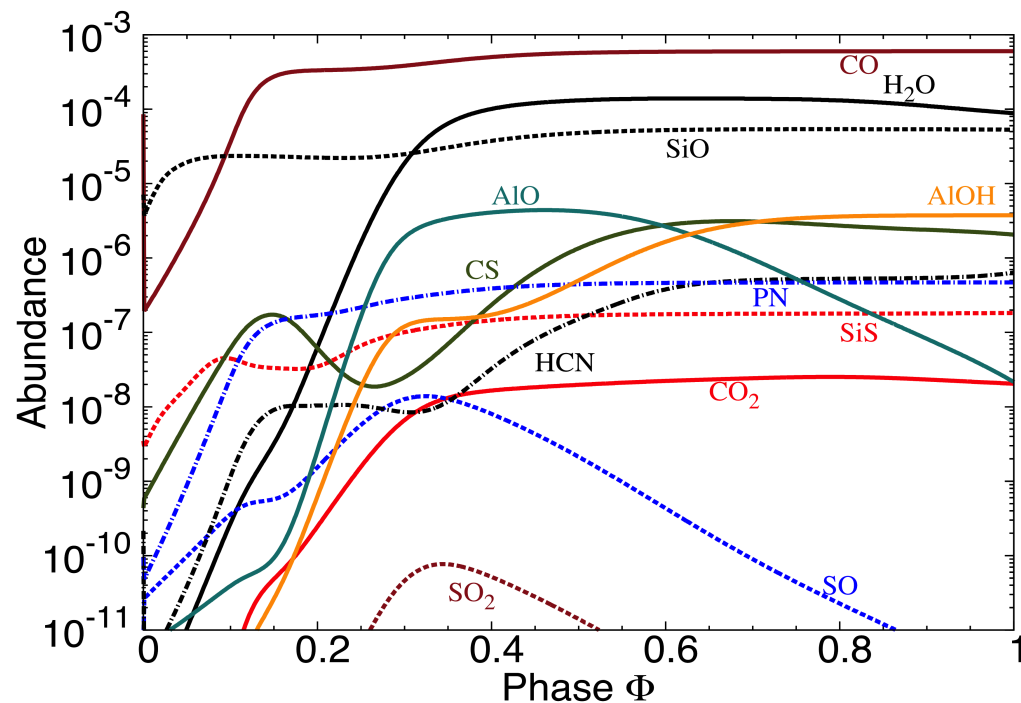
Timescales for dust condensation ?

Consider from hydro models *Bowen 1988, Nowotny 2010*

Dust grains keep growing over several pulsation periods, whereas gas phase molecules reform within one period.

- Drift velocities at $r > 3 R_*$ for silicates – 1.5 km s^{-1} correspond to two pulsations to cover $0.5 R_*$

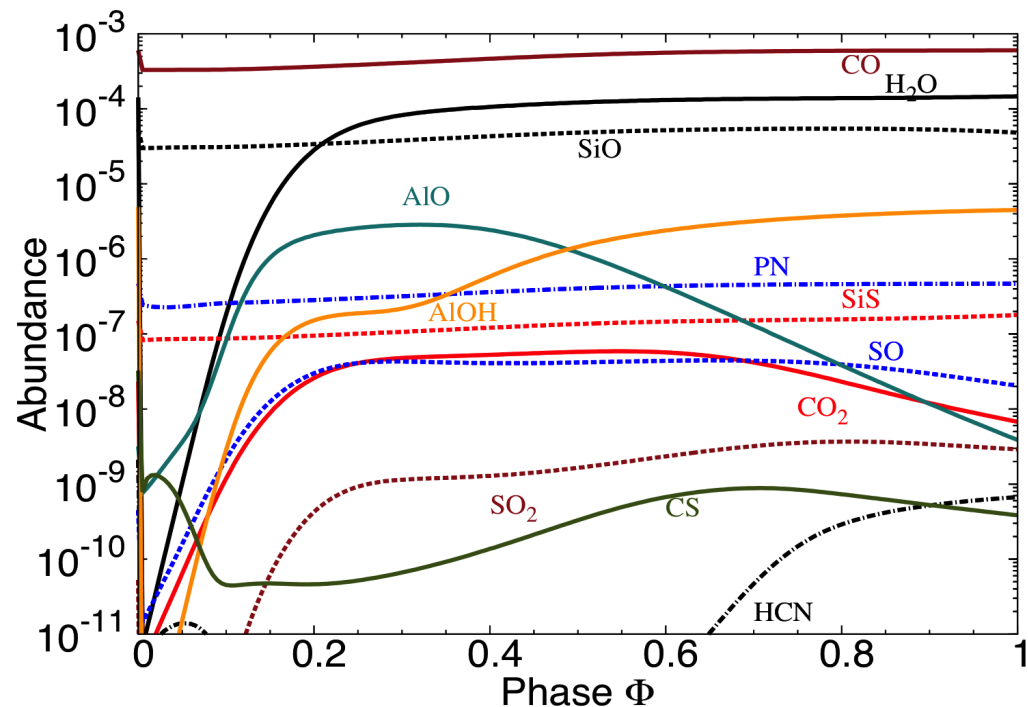
Results: Molecules in IK Tau



1 R_*

CO, H₂O, SiO, SiS, HCl, AIOH & PN form close to the star as soon as gas relaxes and cools down.

Gobrecht 2015 submitted

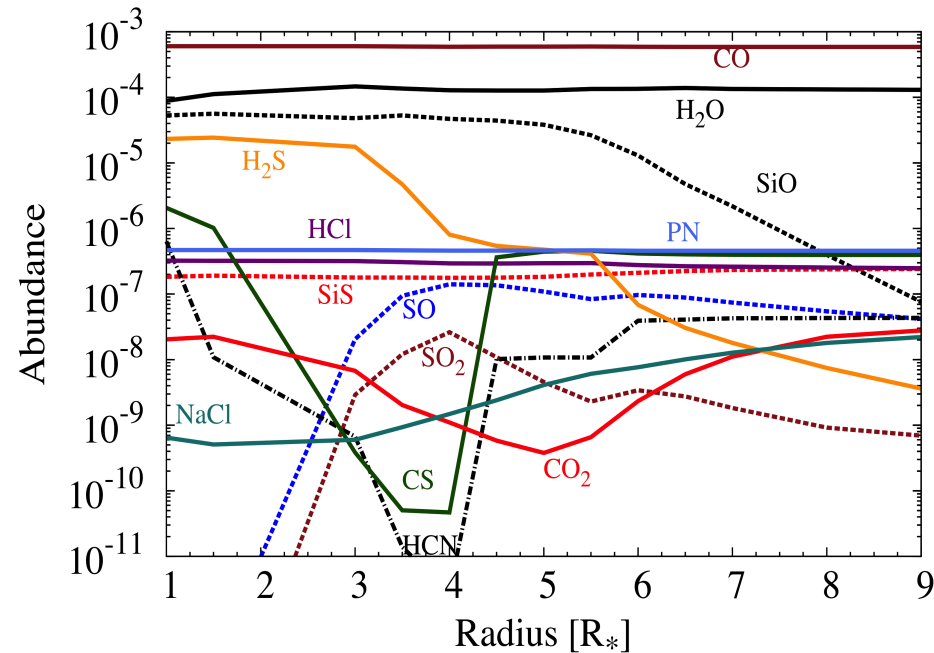


3 R_*

Some molecules are more shock chemistry-dependent. C-bearing species form from CO breaking by shocks

SO, HCN, CS, CO₂

Results: Molecules in IK Tau

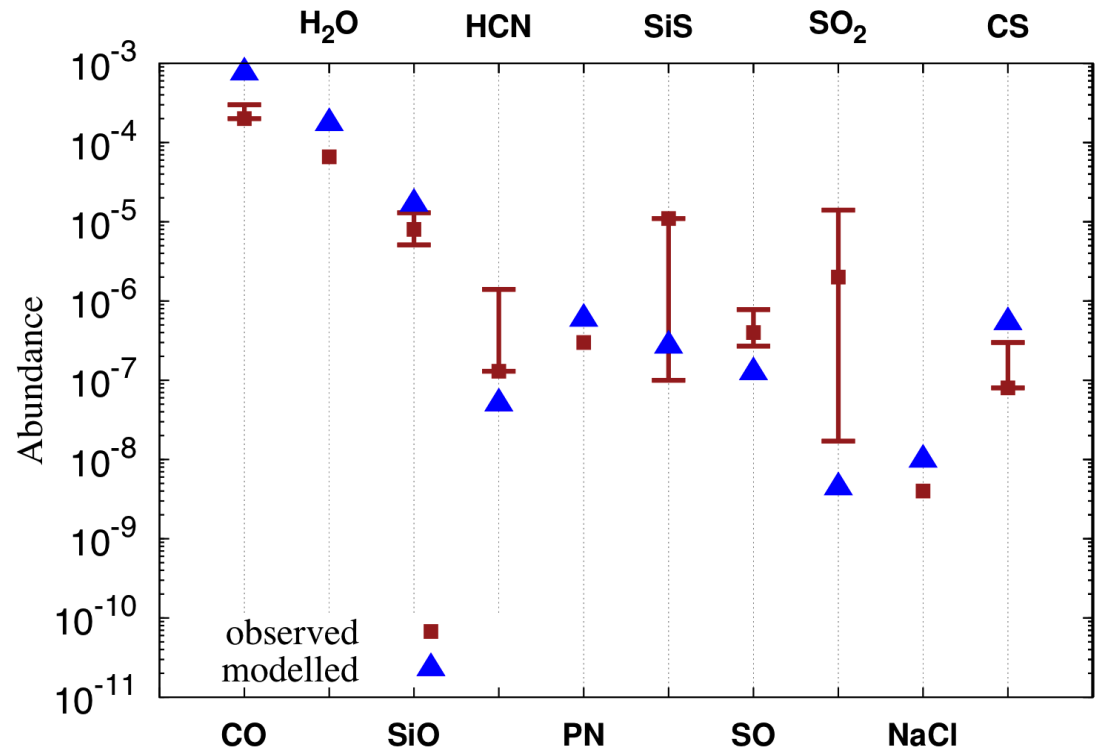


Our parent species include

CO, H₂O, SiO, SiS, PN, SO, HCN,
CS, CO₂, AlOH, TiO, HCl & NaCl

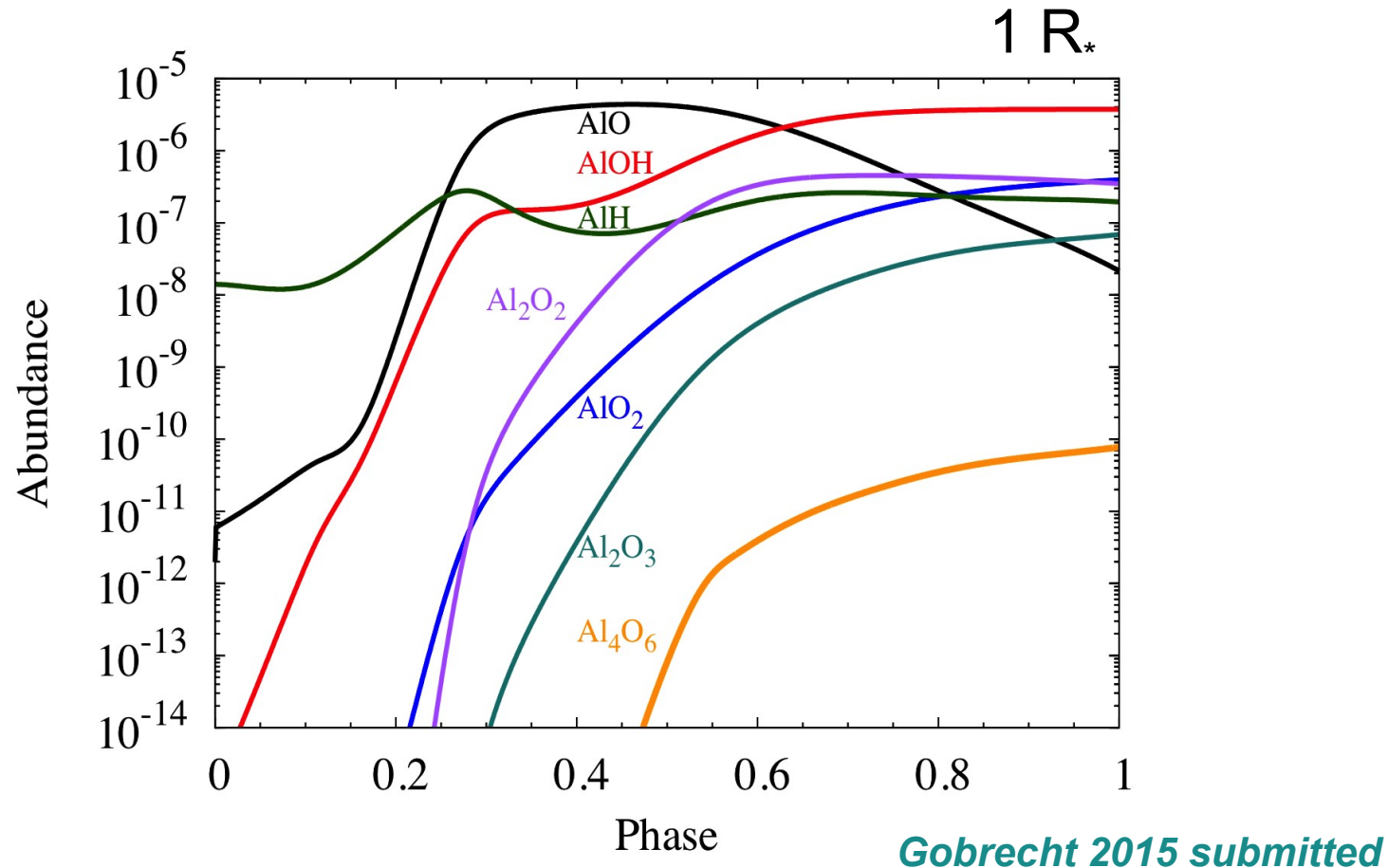
Gobrecht 2015 submitted

- Modelled abundances for 12 molecules at 6 R_* agree well with observations
- Validate shock chemistry scenario → strong impact of shocks on the gas and solid phases of the inner wind
- Discrepancy for SO₂



Results: Dust clusters in IK Tau

Alumina dimers Al_4O_6 :



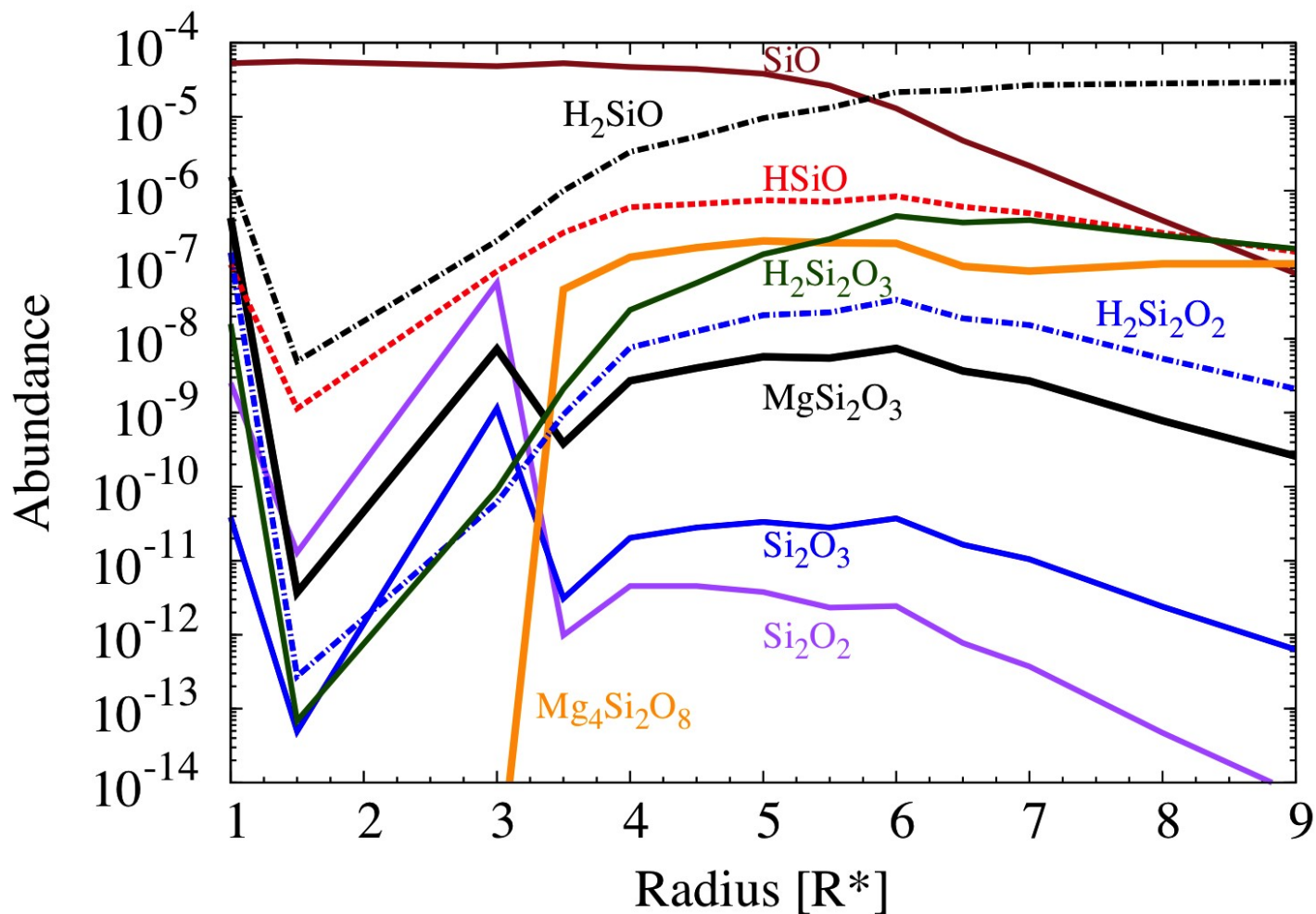
Form at 1 R_* when $T_{\text{gas}} < 2000$ K

Abundance of dimers on the low side

Other formation channels, which deplete AlOH ?

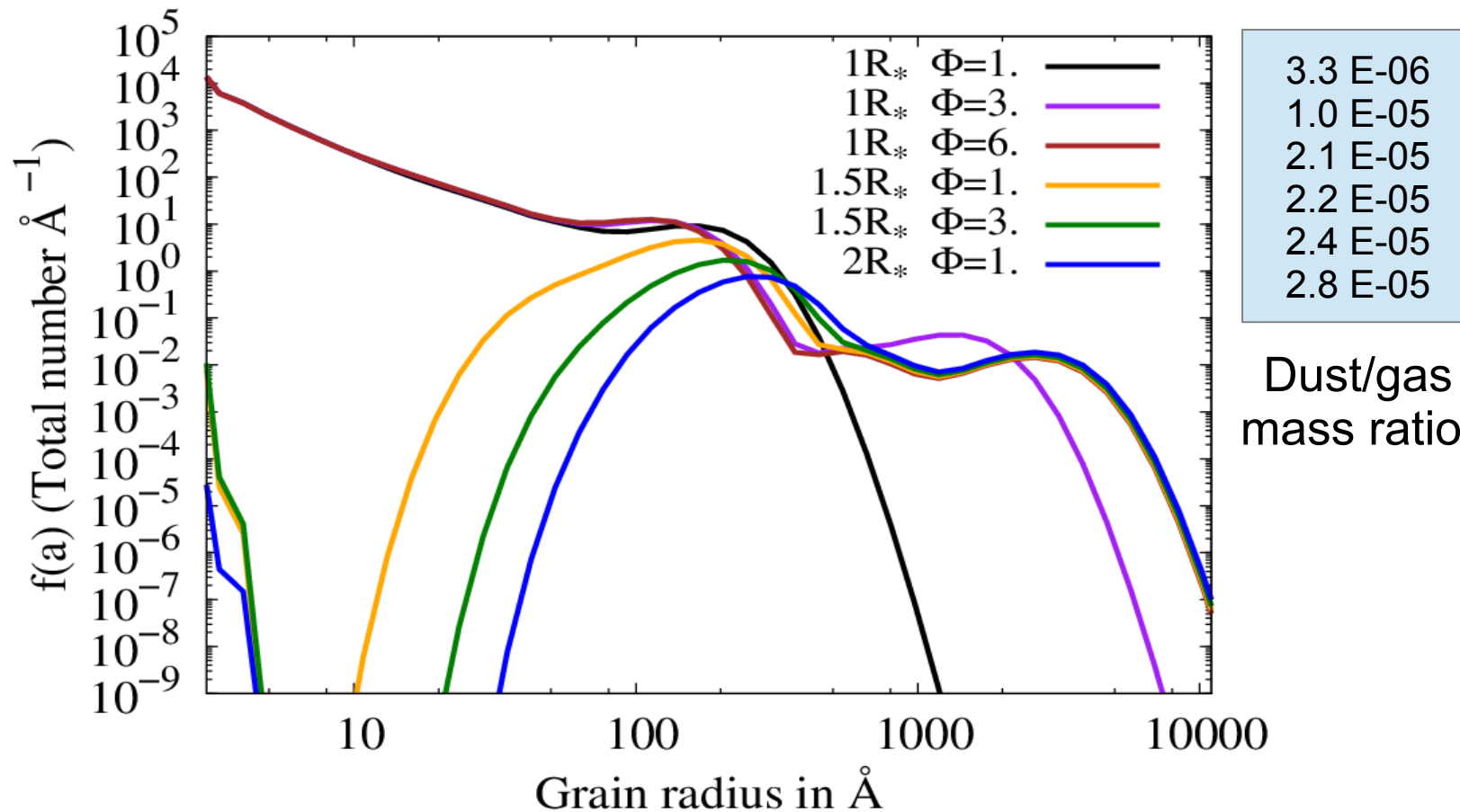
Results: Dust clusters in IK Tau

Silicates: forsterite dimers $\text{Mg}_4\text{Si}_2\text{O}_8$



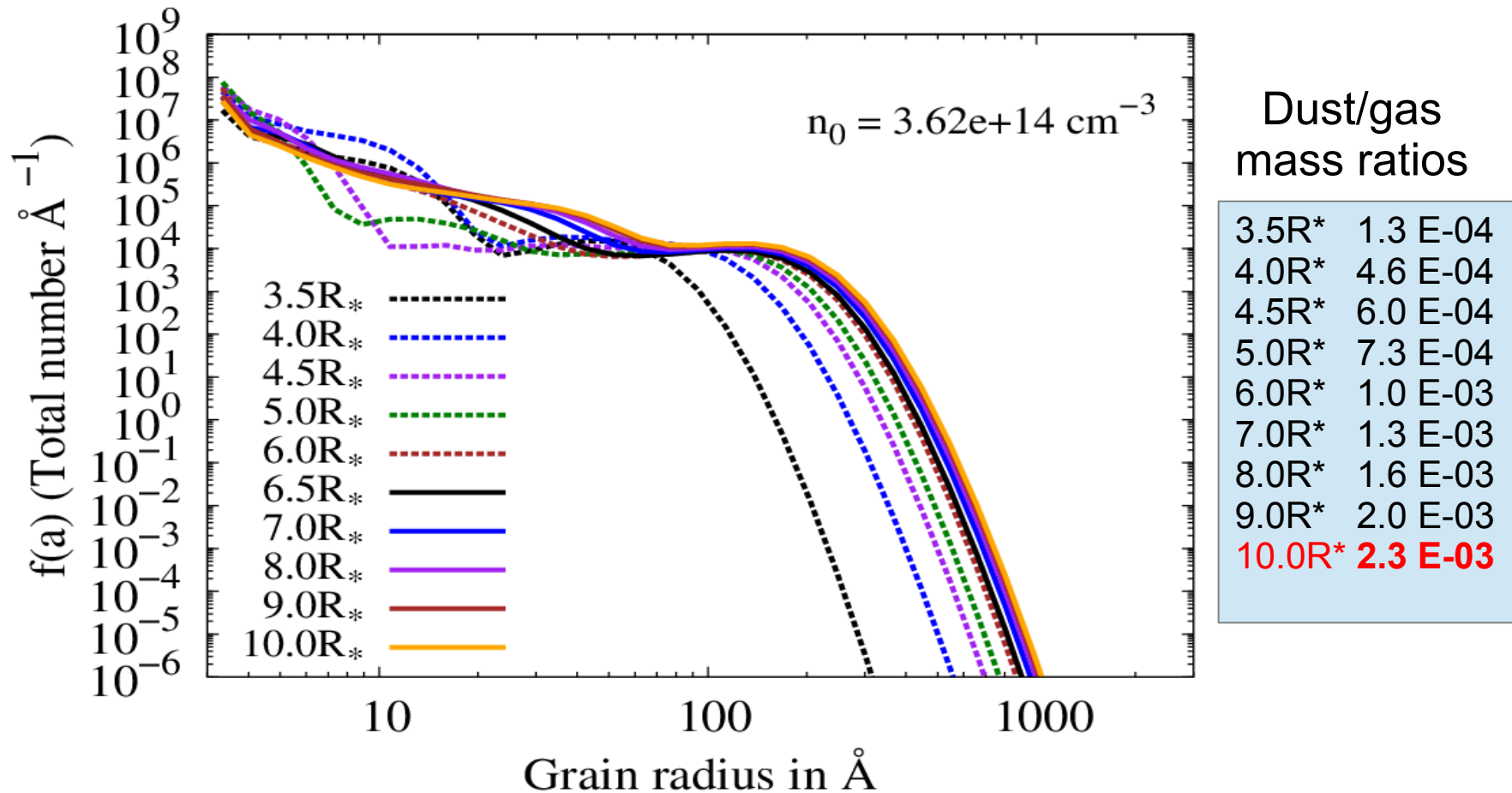
Start forming at $3.5 R_*$ from HSiO dimerisation

Grain size distributions: alumina



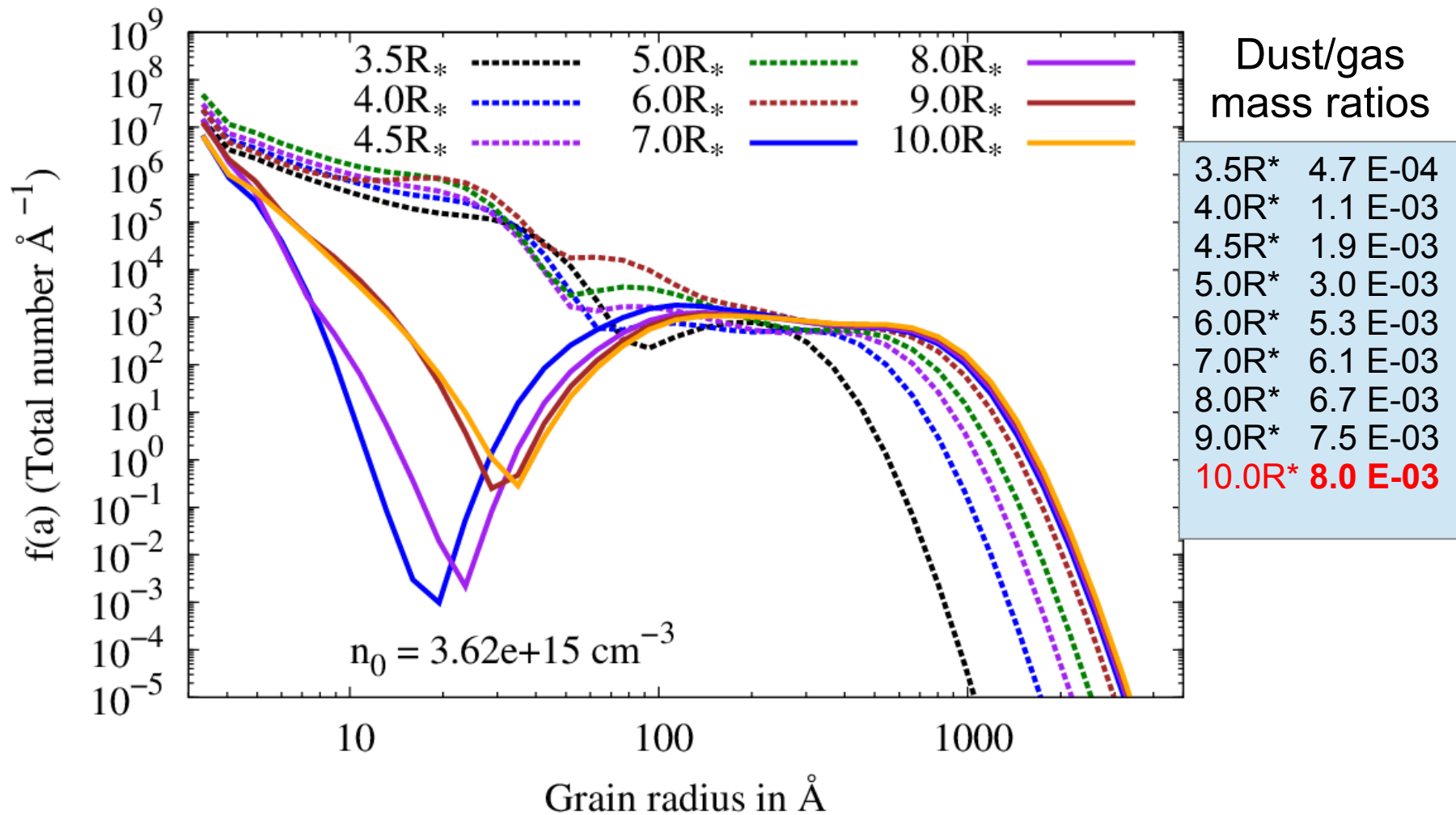
Large grains $> 0.02 \text{ }\mu\text{m}$ are already formed at 1, because gas densities are high in the postshock gas

Grain size distribution: forsterite



- Forsterite grains grow to larger sizes with increasing number of pulsations and radius
- Dust/gas mass ratio after 8 R* agrees with observations
- Grain size peaks at $0.02 \mu\text{m}$, which is a bit low (from obs. $a = 0.1 \mu\text{m}$)

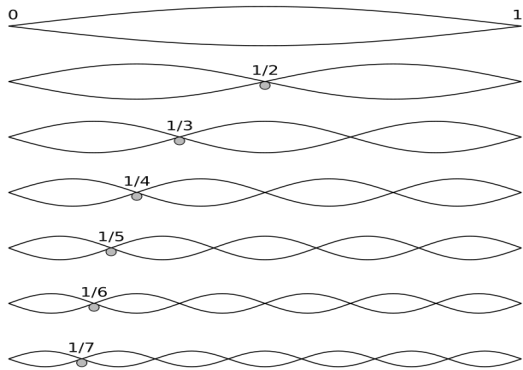
Enhanced density: forsterite



A factor x10 in gas density results in grain size distributions peaking at $\sim 0.1 \mu\text{m}$ → inhomogeneous wind will help!

Semiregular variables

Regularity classes (Mira, Sra, Srb and Irregular) are rather loosely defined in GCVS



Mira: $M_v > 2.5$ mag, $P > 100$ d

Sra: Mixture of Miras and Srbs

Srb: $T_* > 3200$ K, $P < 150$ d

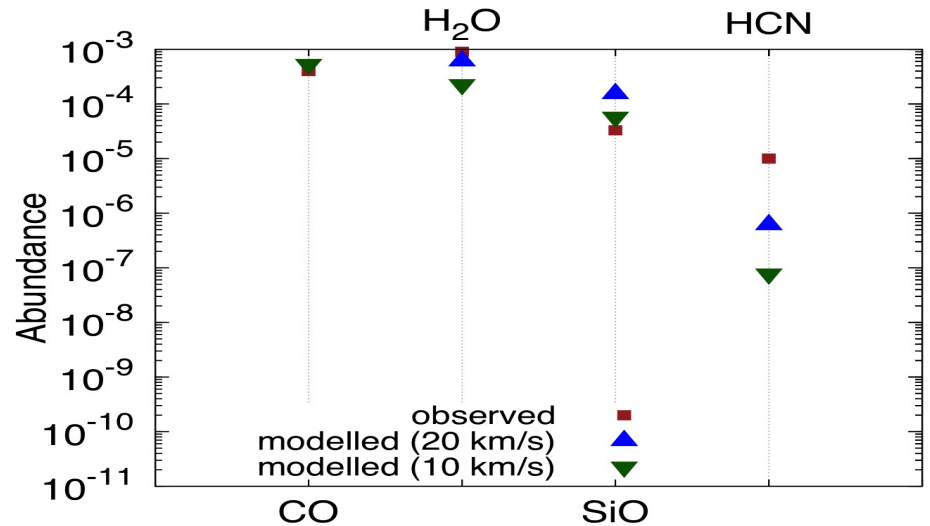
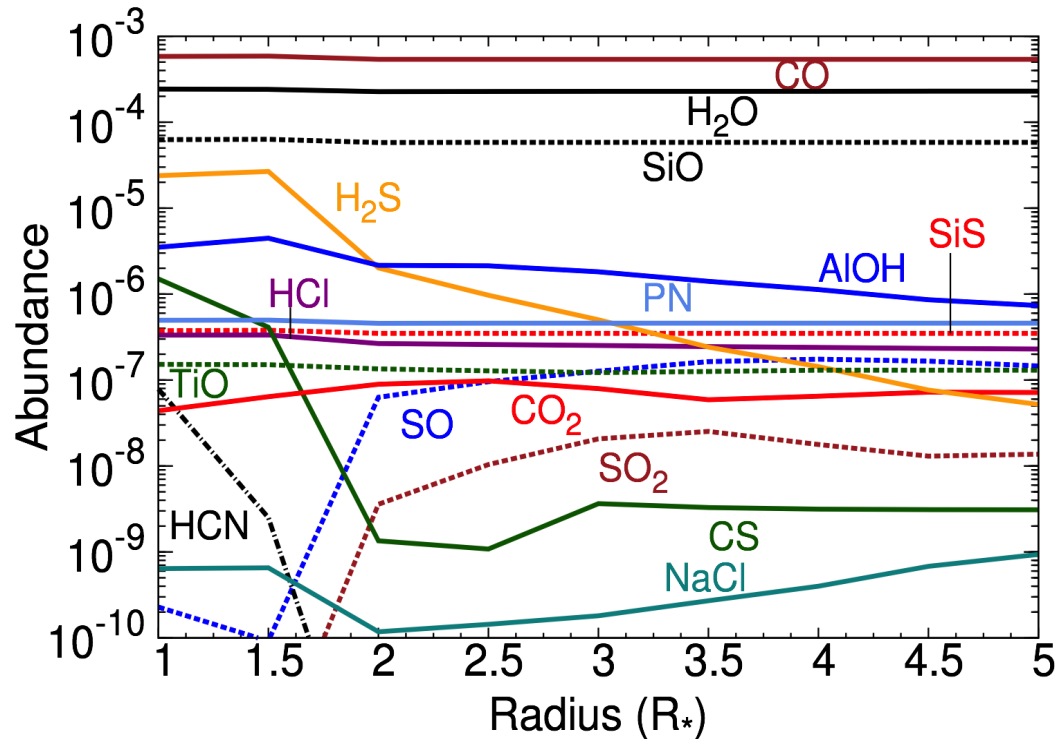
Stellar parameters of SRVs compared to Mira variables:

T_* , M_* are higher

P , R_* , C/O , v_{term} , \dot{M} are lower

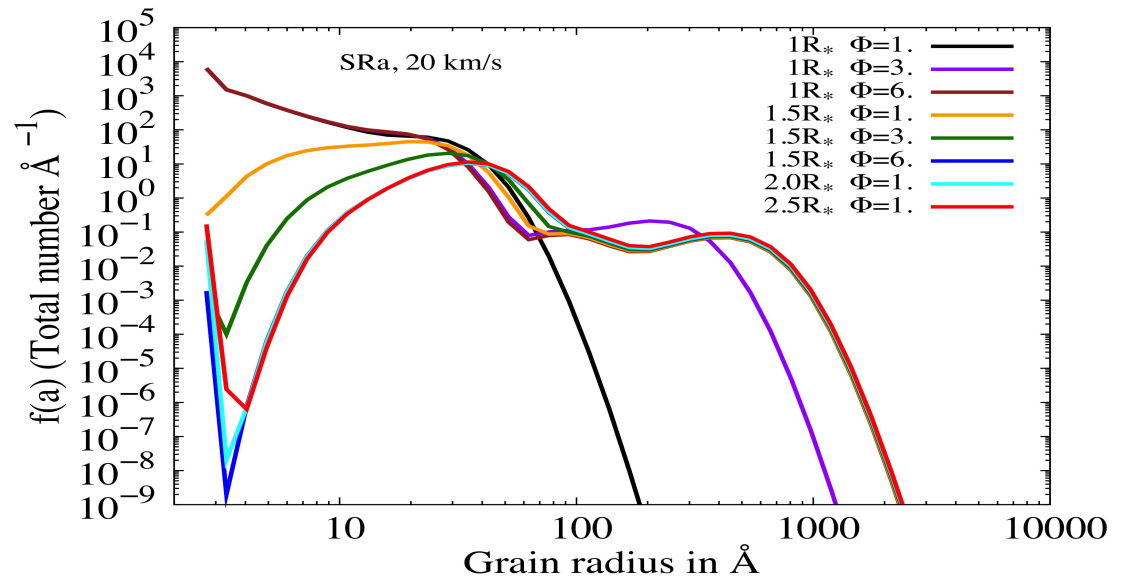
=> interpreted as stars on the early AGB

Results for W Hydrae (SRa)

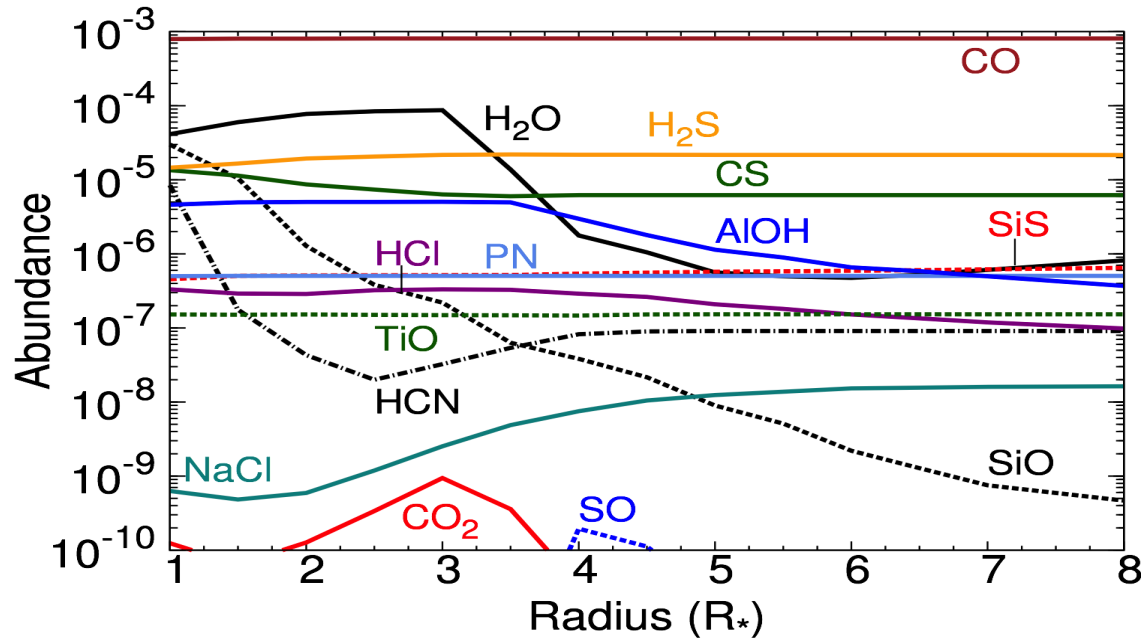


Khouri 2014

Alumina grains form,
Silicate clusters form, but
at too low gas density to
efficiently condense

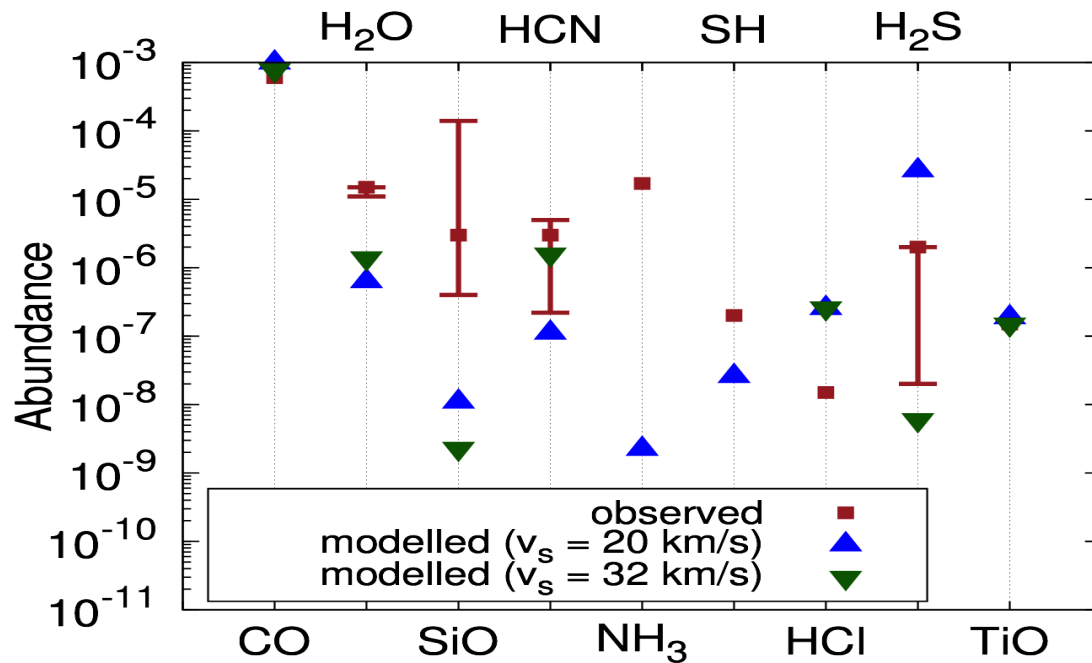


Results: S-type star



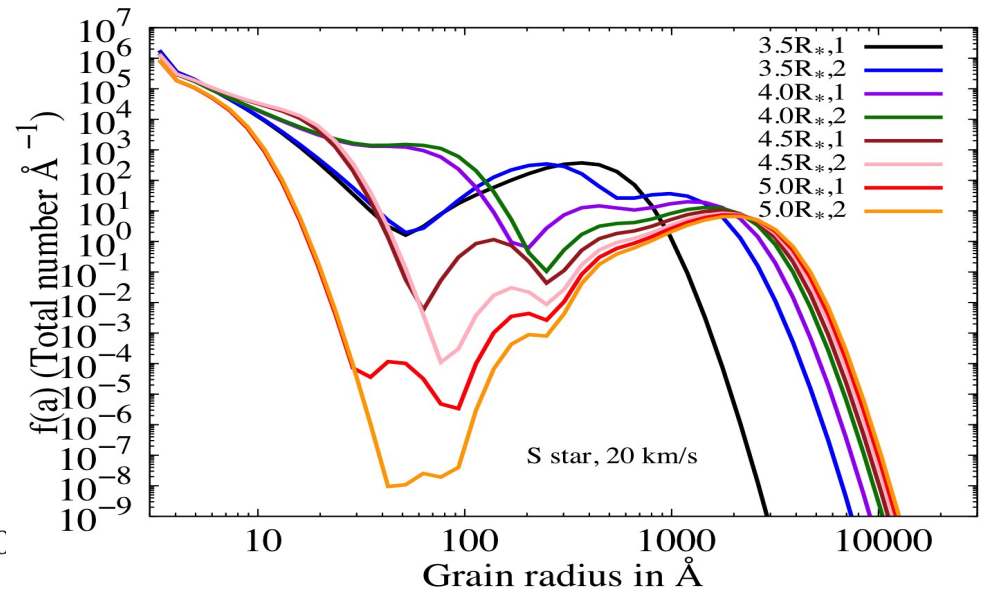
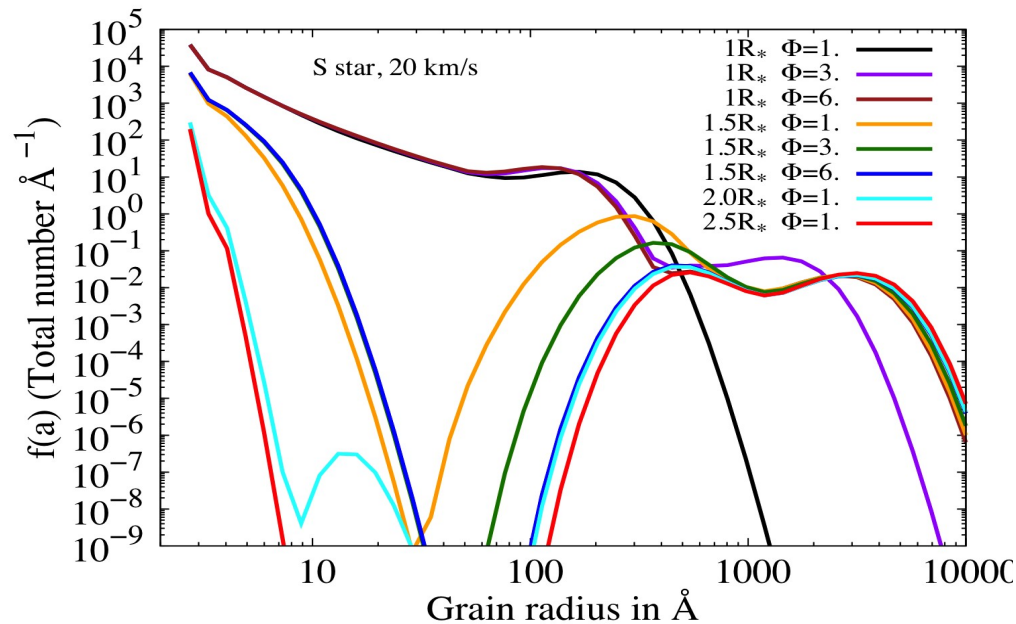
Molecular abundances vs. Radius for a 20 km/s shock and parameters for W Aquilae

Danilovich 2014



Comparison of modelled abundances with the most recent Herschel/HIFI and previous observations

Results: S-type star

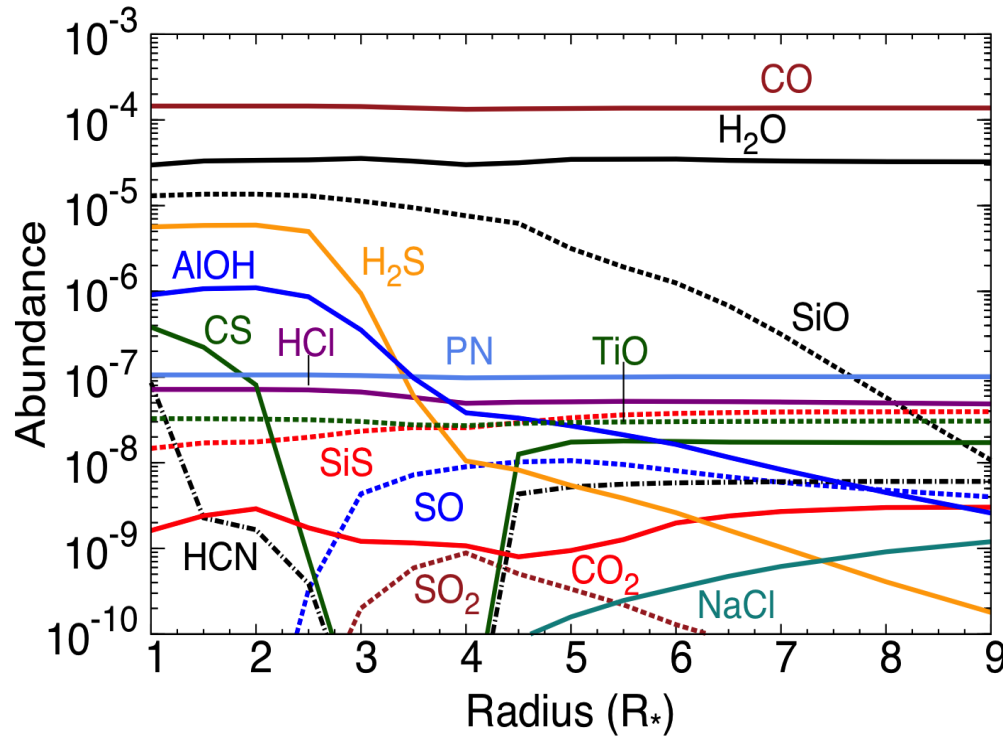


Alumina and forsterite grains form in quantity and large sizes, despite the comparable high C/O = 0.95.

Observations predict a mass loss rate of $3 \cdot 10^{-6} M_{\text{sun}}/\text{yr}$ and a dust-to-gas mass ratio of $5 \cdot 10^{-3}$. This is consistent with our dust-to-gas mass ratio at $8 R_*$:

$\Psi = 4 \cdot 10^{-3}$ for a 20 km/s shock and $\Psi = 6 \cdot 10^{-3}$ for a 32 km/s shock

Miras in the SMC



Alumina dust:

$$\Psi = 1.3 \cdot 10^{-8}$$

$$A_{\text{grain}} = 13 \text{ \AA}$$

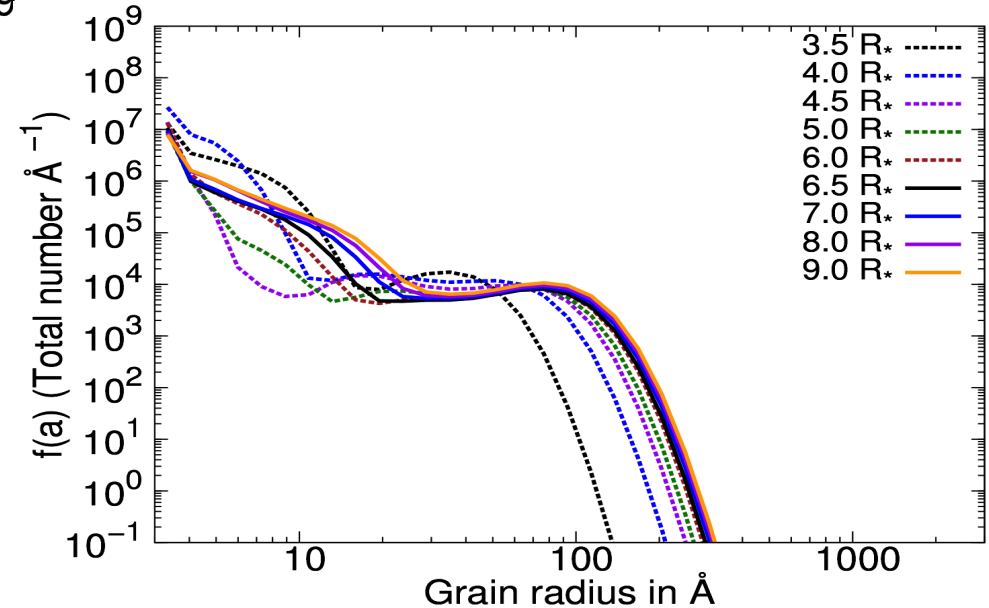
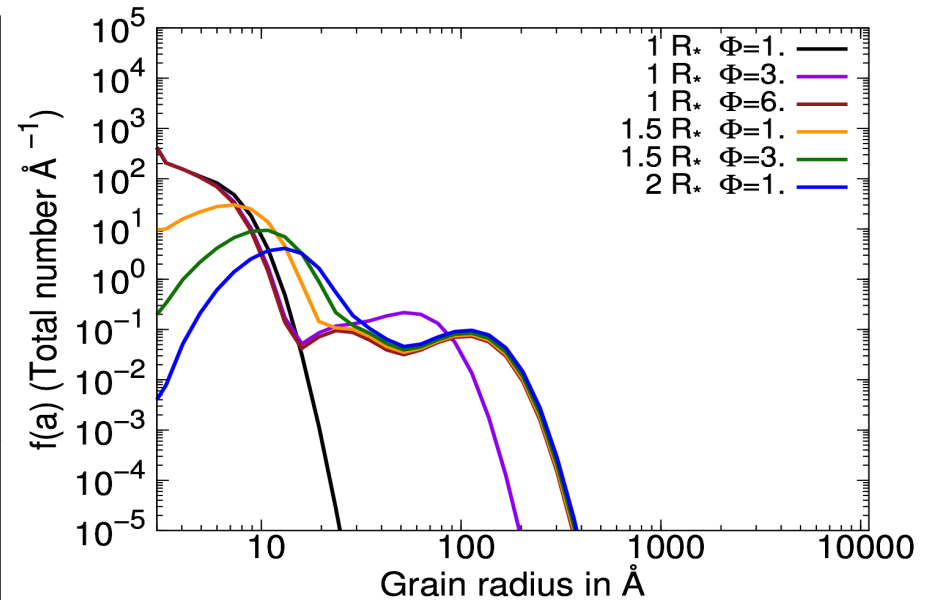
$$Z = 0.003$$

$$D = 61 \text{ kpc}$$

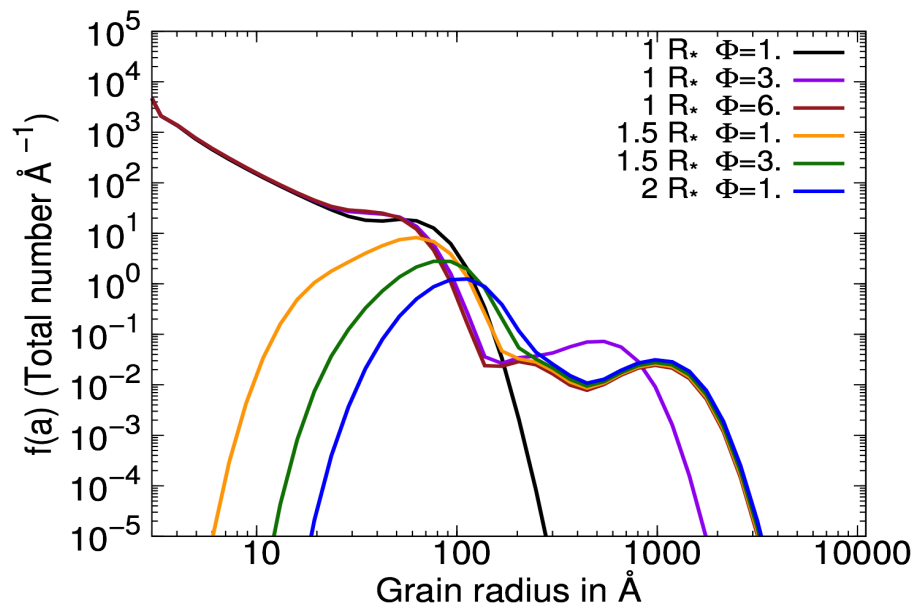
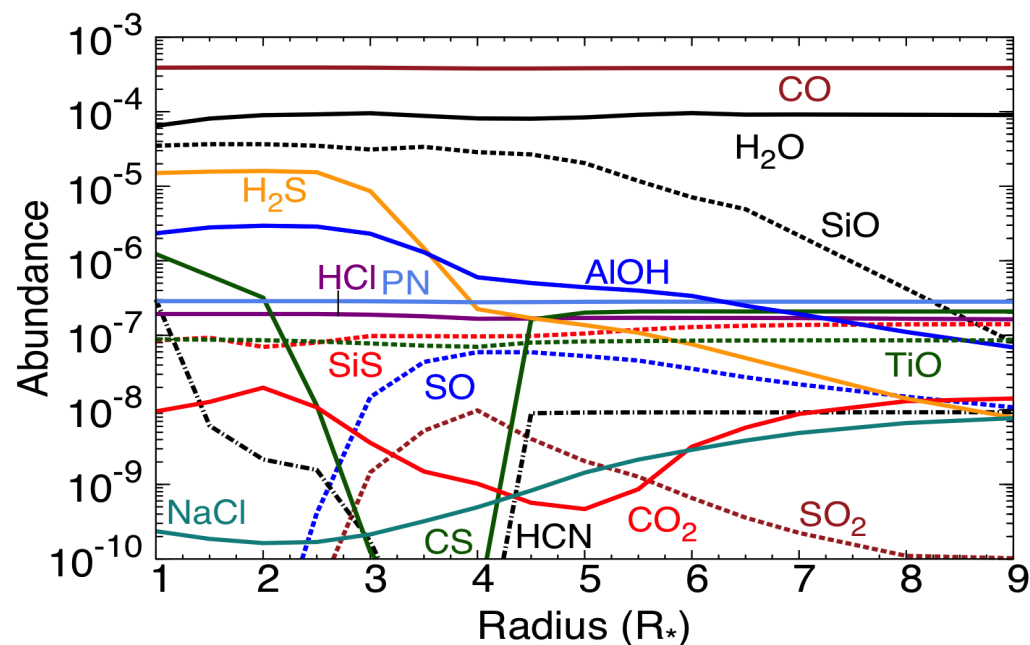
Forsterite dust:

$$\Psi = 3.1 \cdot 10^{-4}$$

$$A_{\text{grain}} = 75 \text{ \AA}$$



Miras in the LMC



Alumina dust:

$$\Psi = 3.2 \cdot 10^{-6}$$

$$A_{\text{grain}} = 115 \text{ \AA}$$

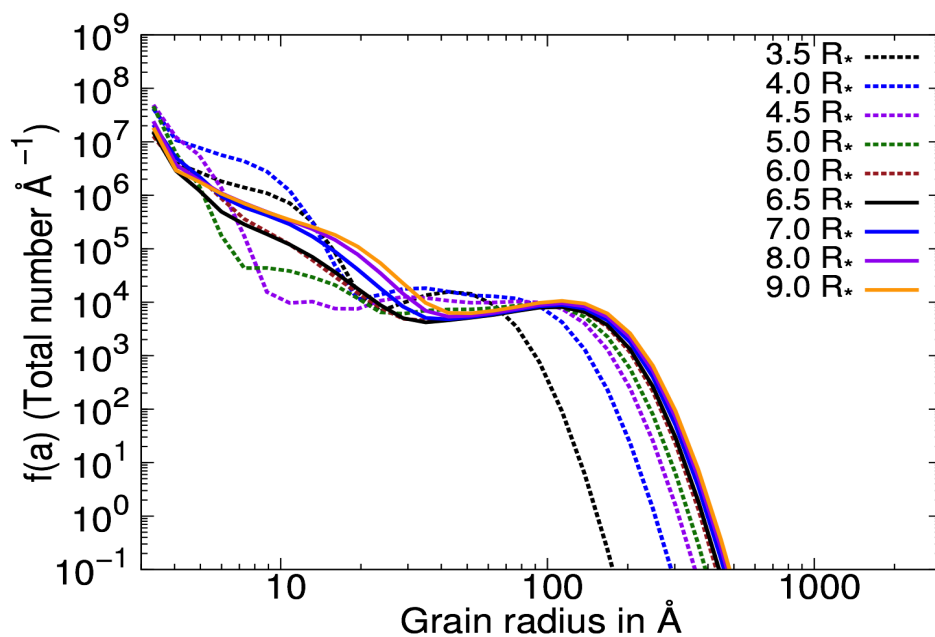
$$Z = 0.008$$

$$D = 50 \text{ kpc}$$

Forsterite dust:

$$\Psi = 1.1 \cdot 10^{-3}$$

$$A_{\text{grain}} = 110 \text{ \AA}$$



Conclusions

- Pulsation models of the inner wind of Mira stars well explain observed molecular abundances of H_2O , OH , SiO , SiS , **CO**, **CO₂**, **HCN**, **CS**, SO , PN and HCl
 - NH_3 , SO_2 and PO cannot not reproduced by the models. Observations indicate that these species
 - are located outside the inner envelope or
 - synthesized by processes not considered here (photo-chemistry, grain surface reaction)
 - Alumina grains ($> 0.1 \mu\text{m}$) form close to the star at $r \leq 1.5 R_*$
 - Silicate (forsterite) grains form between $4 R_*$ and $6 R_*$ from a new nucleation route involving HSiO
- Consistent with recent MIDI/VLT observations

Conclusions

Semi-regular (SRV) model: Molecular abundances agree with observations (CO, SiO, HCN). Alumina dust forms, but silicates can hardly be synthesized, owing to low densities.

S-type star model: Models predict large amounts of alumina and silicate dust. Modelled abundances (in particular SiO and H₂O) agree with observations before the onset of forsterite formation.

Low metallicity model: Smaller amounts of clusters and dust are derived, owing to the lower availability of heavy elements.

For the first time, detailed non-equilibrium chemical models accounting for

- gas phase,**
- cluster nucleation, and**
- dust condensation,**

are set up for the inner winds of AGB stars (O-rich Miras, semi-regular, S-type, and Miras in the SMC/LMC)

and explains the prevalent molecules and dust condensates.

Questions

Any questions?

Suggestions for improvement ?

Looking for experts in Hydrodynamics!